

EXPERIMENTAL STUDY OF UREA GRANULATION

*U. IRSHAD, M.N. SHARIF, F. RABBANI, A. RAUF and M. SALEEM

Institute of Chemical Engineering and Technology, University of the Punjab, Lahore, Pakistan

Urea is a nitrogenous fertilizer available in two commercial forms, prills and granules, in Pakistan. Prills are more common in Pakistan, however, it has some problems associated with it. Those are moisture, biuret contents of urea prills, prill size, hardness of prills, caking of prills and urea dust emission. Due to these problems urea granulation is favoured over prilling. Urea granulation is studied on laboratory scale using pan granulator and effect of different parameters like binders (water and urea solution), quantity of binders, rpm of pan granulator, time of granulation and angle of inclination on granulation yield is studied. Water and urea solutions of different concentrations are used and results reveal that concentration of urea solution is proportional to rate of granulation. A threshold quantity of binder is required for optimum granulation yield. RPM of pan is inversely proportional to rate of granulation. Granulation yield is also proportional to time of granulation however it becomes critical at a certain point. Angle of inclination of pan has no significant effects on granulation yield.

Keywords: Optimum granulation, Rate of granulation, Granulation yield, Pan granulation, Urea granulation, Prilling

1. Introduction

Commercially, fertilizer urea can be purchased as prills or as a granulated material. Depending upon the degree of recycling of the unconverted ammonia and carbon dioxide, the processes of production of urea are, Once through Process, Partial Recycle Process and Total Recycle Process.

Prills are small round aggregates that are artificially prepared. Prilling is a process by which solid particulates are produced from molten urea. Molten urea is sprayed from the top of prill tower. Surface tension causes the liquid to adopt a spherical shape as this result in the smallest surface area to volume ratio. As these droplets fall through a counter-current air flow, they cool and solidify into spherical particles [1].

Prill tower can be classified into two categories on the basis of draft used which are Natural draft Tower and Forced/Induced draft Tower. The prilling tower is a reinforced concrete structure with cylindrical wall. The tower includes the spinning basket room in the tower top and the product scrapper located in the tower bottom. Windows for air inlet and outlet are located on the bottom and the top end of the wall respectively. The evaporation section, the elevator and stairs are

located in a side structure made of steel [2].

Production of prills from molten urea solution involves number of problems that can not only spoil the product quality but in some cases can make the process highly undesirable in aspects of physical and chemical properties of the product. Main problems [2] in the operation of prilling that are common in Urea Industry are moisture, biuret content, prill size, crushing and impact strength, caking of prills and urea dust losses

Selecting a suitable agglomeration technique for urea industry has won wide attention in modern days due to increased population of the world that resulted in increased demand of crops to meet the need. The key question in this regard is what must be the method of agglomeration that caters to the needs of customers and abilities of industries? A dynamic research is vital for comprehensive answer to this. Prills are usually produced by dropping liquid urea from a "prilling tower" while drying the product. The prills formed are smaller and softer substance than other materials commonly used in fertilizer blends. Urea can be manufactured as granules. Granules are larger, harder, and more resistant to moisture. As a result, granulated urea has become a more suitable material for fertilizer blends. Commercial granulation of urea is done by the processes;

* Corresponding author : umar_ce@yahoo.com

Fluidized Bed Granulation, Spouted Bed Granulation [3].

The manufacture of prills is rapidly decreasing owing to both environmental problems and product quality as compared to granules. Generally, both crushing and impact strength of the prill is much less than for granule. This cause many problems in handling both at the plant and in shipping. [4]. Granulation is more popular than prilling in those countries that are very much particular about their environment. There are two granulation methods, drum granulation and pan granulation [5]. In drum granulation, solid are built up in layers on a seed granules in a rotating drum granulator/cooler approximately 14ft in diameter. Pan granulator also forms the product in a layering process, but different equipment is used [5].

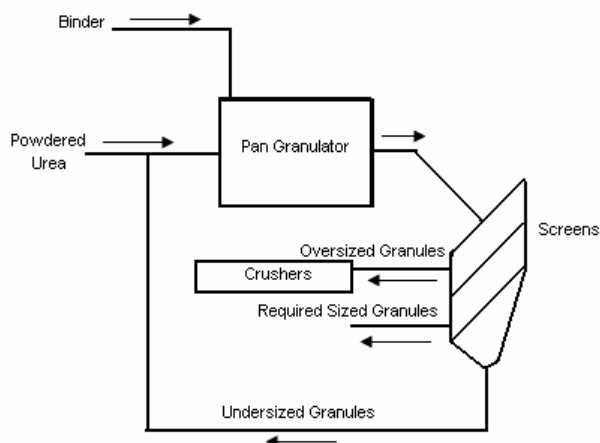


Figure 1. Schematics of the granulation.

The Hydro-Agri process is used in the majority of newly installed plants that are producing granules in U.S.A. this process was developed by NSM of Holland many years ago. They have no plant size limitation and design a single-train unit for production over 3000MTD. The C&I Girdler drum system has been very successful, but cannot compete in today's market because of restrictions in the train size. Toyo has successfully developed a spout-fluid technology. Stamicarbon also working on a fluid-bed large-scale single-train plant that is somewhat similar to the Hydro-Agri design [4].

Urea granulation is studied experimentally in this work. Factors affecting granulation are studied such as rpm granulator, time of granulation, concentration of binder used, amount of binder used and angle of inclination.

2. Experimental Work

Granulation of urea was done in the laboratory using pan granulator. The granulator consists of a rotating pan with adjustable speed motor. The pan diameter is 46.5cm and depth is 16cm. The pan can hold upto 1.5 kg of charge depending upon the density of charge. Spray nozzle distributes fine mist of binder. Powder and solution flow continuously at predefined rate. The slope degree of its pan can be adjusted from 40° to 55°.

The pan revolves at a certain angle with the horizontal plane driven by the motor. The powder will rise along with the revolving pan under the friction between the powder and the pan, on the other hand, the powder will fall down under the function of its gravity. At the same time, the powder moves to the pan edge because of the centrifugal force. The powder material rolls in a certain trace under the function of these three forces. It gradually becomes the required size [4]. Product is withdrawn and classified using standard sieves (US ASTM MESH SCREENS) and oversized is sent to crusher and undersized is recycled. Bulk density of the product of required size is determined. Standard deviation of whole product and granulation yield are also calculated.

3. Results and Discussions

3.1. Effect of RPM of pang granulator

Effect of rpm of pan granulator has a very pronounced effect on the amount of urea granulated as shown by Figure 2. Because as the speed of the pan increases the amount of granules formed decreases greatly. During the course of granulation, collision between granules themselves, also with the wall of the pan, and with doctor knife take place. At high speed of pan i.e. more rpm of the pan collision becomes sever and it breaks the granules. At low speed of pan outer most granules loose contact with the wall of the pan depend on the balance between gravitational and centrifugal forces and cause breakage of themselves and other granules as well. Both bulk density and granulation yield show increasing trend up to 25rpm and then decrease. Thus granulation yield is maximum at 25rpm. Maximum also is the bulk density. Hence, the storage volume required will be lower. Consequently the optimum speed of pan granulator emerges out to be at a value of 25 rpm.

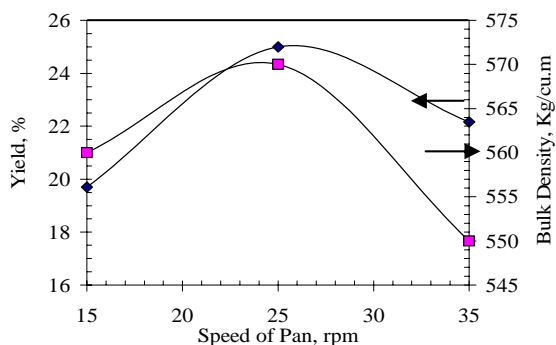


Figure 2. Effect of speed of pan on yield and bulk density.

Effect of rpm of pan on mean particle diameter and standard deviation is shown in Figure 3. Mean particle diameter increases with increase in rpm of pan granulator up to 25rpm. Beyond this speed, particle diameter goes decreasing. Standard deviation is also minimum at 25rpm thus converging the focus of optimality on the speed.

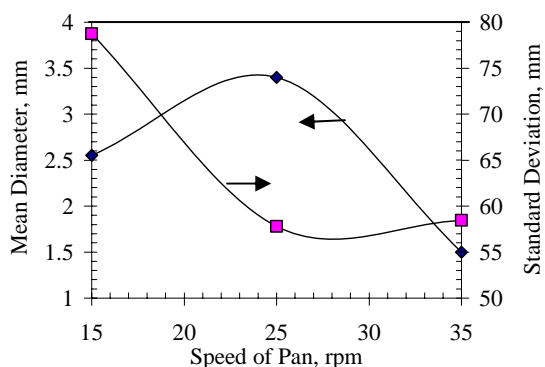


Figure 3: Effect of speed of pan on mean diameter (d_{50}) and standard deviation.

So, there exists a certain critical speed of pan which is favorable for the granulation of urea. Thus the product obtained falls in a narrower range of size and more product of required size is obtained at this speed.

3.2. Effect of time of granulation

Effect of time of granulation on yield (%) and bulk density is shown in Figure 4. Bulk density first decreases and beyond 30 min. granulation time, it shows ascendancy in trend. An increase in the bulk density with increase in granulation time supports the fact that more desirable product is formed. Thus quality of the product is higher. Meanwhile the granulation yield (%) also goes on

increasing. Thus both quality as well as %yield are improved but half hour time is optimum because of minimum standard deviation and acceptable mean particle diameter (d_{50}).

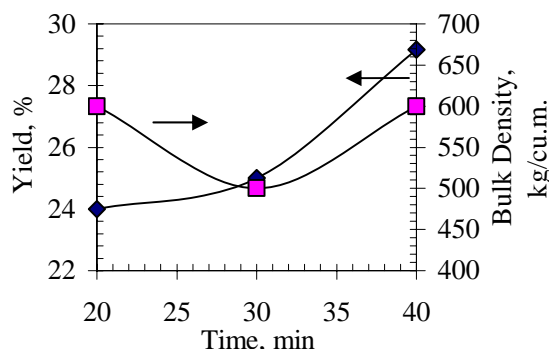


Figure 4. Effect of time of granulation over granulation yield and bulk density.

Effect of time of granulation on mean particle diameter and standard deviation is shown in Figure 5. By increasing the time of granulation, mean diameter of the product (granules) increases. At the same instant, the standard deviation is much less which indicates optimum granulation time is 30 min. When granulation time is increased beyond 30 min, standard deviation shows optimality on half hour granulation period.

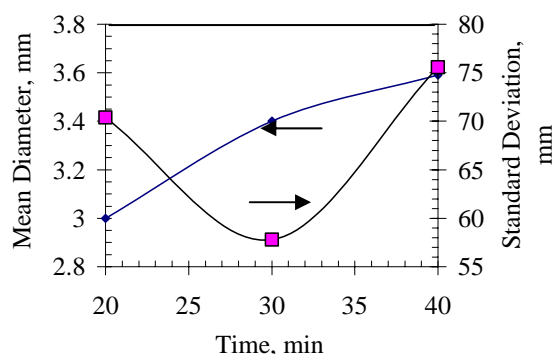


Figure 5. Effect of time of granulation on mean diameter (d_{50}) and standard deviation.

3.3. Effect of concentration of binder on granulation

Different chemicals that can be used for binding during granulation are water, urea solutions of different concentrations, molten urea and polyurethane. Because of high melting point of urea (135°C) it can not be used as it readily

solidifies at ambient temperature. It has been evident from the experiments that operation using water gives much less granules than using urea solution. Also as the concentration of the urea solution rises, the quantity of urea granules increases depending upon the binding strength of urea solutions increases with increase in urea concentration. Effect of concentration of binder on granulation yield and bulk density is shown in Figure 6. The %yield is maximum at 60% concentration of binder. The bulk density also shows on increase after enhancing concentration of binder emerges to be 60% .

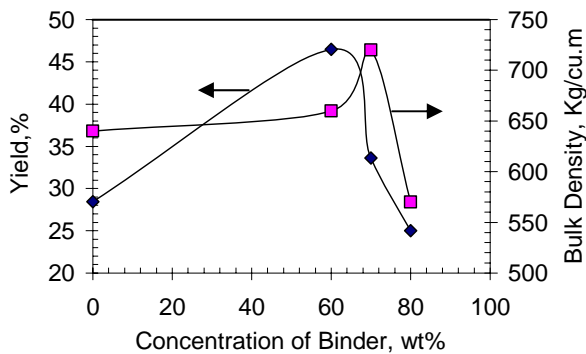


Figure 6. Effect of concentration of binder on yield and bulk density.

Standard deviation increases while concentration of binder is increased from 0 to 60% but increasing trend is not sharp as shown in Figure 7. The mean particle diameter is also acceptable at this concentration of binder. Mean diameter is increasing smoothly till 60% concentration of binder but at 70% concentration it sharply falls. Then it shoots up for 80% concentration that indicates that result on 70% concentration does not represent the actual trend.

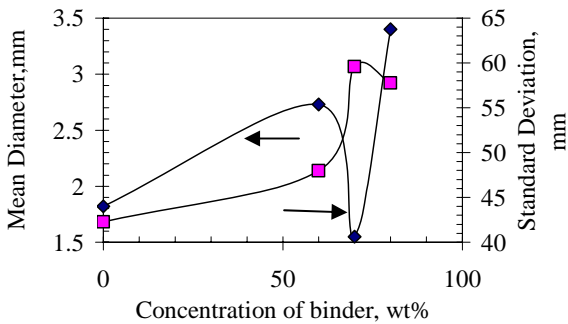


Figure 7. Effect of concentration of urea solution on efficiency of granulation.

3.4. Effect of amount of binder

As for as the quantity of chemical used i.e. water and urea solution of different concentrations is concerned it is evident from the experiments that initially with the increase of the quantity of binder used, the amount of urea agglomerated or granulated increased but with more increase, in the quantity it showed the reverse effect. It means a definite proportion of binder and urea powder is needed for granulation. Because when quantity increases it yields very soft granules that break during the process. Though bulk density is not maximum at 100gm amount but % granulation yield is maximum as shown in Figure 8.

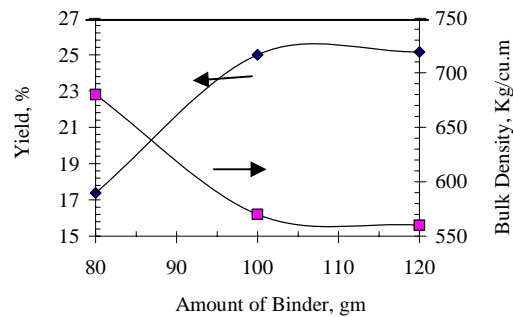


Figure 8. Effect of amount of binder on yield and bulk density.

Mean particle diameter increases when amount of binder is increased up to 100g and then decreases beyond that amount as shown in figure 9. At 100g amount of binder, standard deviation is also minimum reflecting optimality of 100gm amount of binder. Consequently this amount of binder (100gm) comes out to be optimum.

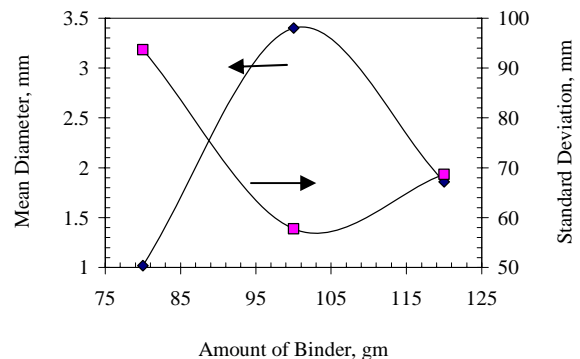


Figure 9. Effect of amount of binder used on mean diameter (d_{50}) and standard deviation.

3.5. Effect of inclination of pan

Experiments were performed for different inclinations of the pan within the operational constraints of the equipment (40° - 55°) but there is no significant effect observed on the granulation yield.

4. Conclusion and Outlook

It may be concluded from this experimental study that optimum conditions of granulation using laboratory pan granulator are:

Time of granulation: 30 min,

Concentration of binder: 60%,

Amount of binder: 100gm,

Speed of pan granulator: 25rpm.

Liquid urea is no longer in current use due to volatile nature of ammonia. Most of fertilizer manufacturing plants working in Pakistan are producing prills. The only plant that is producing urea granules instead of prills is Fauji Fertilizer Bin Qasim Limited. However, granulation technique is intended to be more suitable recently and will replace prilling method in future. Future researchers will have to concentrate mainly on region of improvement and further progress in granulation of urea. Restricting a particular group of constraints, the advantages can be sought for and traditionally these will be welcome in manufacturing as well as marketing sector. The future of urea granulation demands business like organization of experimental studies and model attitude in order to make it an activity that provides best service to agriculture and hence to agri-based economies of the world.

Acknowledgement

Authors acknowledge the support provided by Institute of Chemical Engineering and Technology for performing the experimental work. Thanks to Mr. Kamil who worked as research assistant.

References

[1] M. E. Fayed and L. Otten, Handbook of Powder Science and Technology, Van Nostrand Reinhold Company, New York (1984).

[2] U. Irshad et al., A Process Report on Urea Prilling and Granulation, I.C.E.T, Punjab University, Lahore (2008).

[3] Fertilizer Manual, United Nations, New York, (1968) pp.105-110.

[4] K. Othmer, Encyclopedia of Chemical Technology, 4th Edn., Volume Supplement, John Wiley and Sons, New York (1998) pp. 597-620.

[5] J.J. Makita and Guy E. Dismantle, Encyclopedia of Chemical Processing and Design, Vol. 60, Marcel Dekker Inc. New York (1997) pp.176-182.

[6] <http://www.en.wikipedia.org/wiki/urea>

[7] http://www.mahii.cn/images/products/pan_granulator_gif.htm.